# The use of an odour wheel classification for the evaluation of human health risk criteria for compost facilities

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**Abstract** Odorants are released during the decomposition of organic waste at compost treatment plants. Composting releases volatile organic chemicals (VOCs), including alcohols, aldehydes, volatile fatty acids, ammonia and other nitrogen compounds, xenobiotic solvents, and various sulphur compounds into the environment as categorised by a compost odor wheel. Each odorant possesses a characteristic odour signature – quality and threshold as well as a toxicity value. This paper presents data relating the human odour detection limit to human health threshold criteria developed by the National Institute for Occupational Safety and Health, Occupational Safety and Health Administration, the United States Environmental Protection Agency Region 9 and the World Health Organisation. This comparison indicates that: (1) the human odour threshold concentrations (OTC) for most compost odorants are far lower than their respective human health risk (regulatory) threshold values, (2) several compost odorants have OTC that are below some of their respective regulatory thresholds and above others (i.e. dimethyl amine, formic acid acetone, ethyl benzene and toluene) and (3) only the VOCs probably present as contaminants in the raw composting material have OTC greater than all of its regulatory thresholds (i.e. benzene). Benzene is the most hazardous VOC associated with composting and should be monitored.

Keywords Compost odour; compost odour wheel; compost risk assessment; odour risk analysis

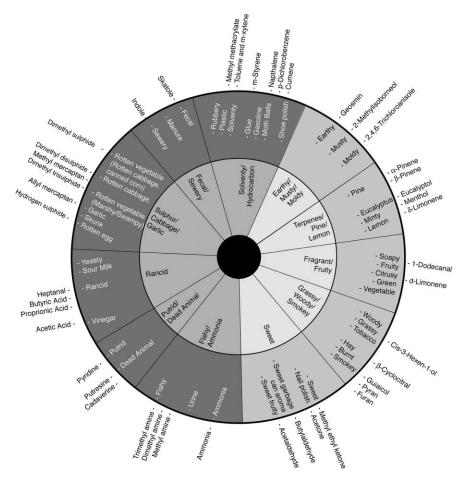
#### Introduction

Odour control at the fence line of composting facilities is a common goal. When odour plumes cross the fence line of facilities, neighbours and community members become alarmed and have human health concerns. This paper evaluates the human detection limits of common compost odorants to their respective human health threshold criteria developed by the National Institute for Occupational Safety and Heath (NIOSH), Occupational Safety and Health Administration (OSHA), United States Environmental Protection Agency Region 9 (EPA) and the World Health Organisation (WHO). This paper will also evaluate odour threshold concentrations (OTCs) and odour characteristics from compost treatment processes. A conservative compost odour wheel is presented to sum up the consensus odour evaluation (Rosenfeld *et al.*, 2004; Suffet and Rosenfeld, 2006). The odour wheel takes into account sulphur, nitrogen and oxygen odorous compounds which are the basis for most of the nuisance odours. Ten of the odorants categories are from the raw material or produced during the composting process. The last odour wheel category includes a set of xenobiotic compounds present probably in the raw material, such as the substituted benzenes (Komilis *et al.*, 2004).

The odorants released from a compost treatment plant vary depending on the raw compost entering the plant and the operation of the treatment processes. The odorants released from compost plants are typically from organic waste which are microbially degraded in sequential order where the sugars are consumed first, followed by cellulose, proteins, lignins, oils and finally, waxes (Killham, 1994). The oxidation state of the odorous nitrogen and sulphur compounds depends on how the compost treatment facility is operating under aerobic or anaerobic conditions. Also, anaerobic conditions are responsible for production of odorous volatile fatty acids.

Human health risk evaluation criteria include: OTC meaning the minimum concentration that humans can sense in the presence of an odorous gas, recommended exposure limits (REL), short-term exposure limits (STEL), preliminary remediation goals (PRG) and permissible exposure limits (PEL). Each exposure limit is established for a particular compound over a particular time period, resulting in an effective dose for an individual. The REL is usually based upon exposure to the compound over an 8–10 hour time period. The STEL is based upon a 15 minute exposure limit. The PEL is based upon a concentration over an average eight hour workday or 40 hour work week to which most workers can be exposed without experiencing adverse effects. Substances at or above PRG concentrations are subject to remediation (OSHA, 1993, 1997; WHO, 2000; EPA, 2004; NIOSH, 2004).

The odour wheel describes the sensory and chemical links between odour descriptors (Figure 1) (Rosenfeld *et al.*, 2004; Suffet and Rosenfeld, 2006). For the compost odour wheel, the odour descriptors have been linked to known compost chemicals as well as potential compost chemicals that have a particular odour but have not been associated



with a particular compost odour source. When compost air odour control methods and sources are linked to specific chemical causes and their odour quality identified, air odour problems will be more easily solved. With this in mind, a compost odour classification wheel was developed as shown in Figure 1 (Rosenfeld *et al.*, 2004; Suffet and Rosenfeld, 2006). The classification scheme serves as a basis for sensory testing and odour control. The 11 categories of odour descriptors are listed and potential chemical causes are reported based upon Table 1A and B. The intensity of each odour quality was not determined and remains the work of an odour panel to refine the evaluations in the future. In this investigation, the "warning" property of odour is evaluated. "Warnings" would be given when a compound's OTC was less than regulatory levels. In this case, there are five possibilities: (1) there is an OTC, but no regulatory limit; (2) the OTC is less than the regulatory limit; (3) the OTC is equal to the regulatory limit; (4) the OTC is greater than the regulatory limit; (5) there is no OTC known.

## Method

Chemical characteristics, OTCs and health risk evaluation criteria for odourous gasses associated with compost are compiled in Table 1A and B. Odorous gasses resulting from composting can include sulphur compounds, nitrogen compounds, volatile fatty acids and other volatile organic compounds. The toxicity threshold values or regulatory limits published by NIOSH, OSHA, EPA and WHO are presented to allow for comparison to the OTC. Investigators do not always give the same name to the odour characteristics, but the consensus descriptions are within a specific type. OTCs are sometimes reported with order of magnitude differences. VOCs such as substituted benzenes that have been shown to be present in compost emissions (Komilis *et al.*, 2004), but probably not associated with odorants produced by the composting process are studied to assess any source of problems.

### **Results and discussion**

#### Sulphur compounds: sulphur/cabbage/garlic

Anoxic conditions at compost treatment plants produce sulphur-type odours (Banwart and Bremner, 1975). Table 1A and B indicate the present knowledge of sulphur compounds and other compounds thought to be present in compost treatment plants that could be part of the compost plant's odour quality. However, the relationship between the OTC of the chemicals present, their relative concentrations and the type of odour has not yet clearly been defined (Rosenfeld and Suffet, 2004).

The OTCs for sulphur compounds are below their respective regulatory limits, effectively giving a warning to exposed individuals. Sulphur compounds' OTCs (Table 1A and B) are generally lower than most other compounds (Ruth, 1986). Hydrogen sulphide has an OTC of  $0.70 \,\mu g/m^3$  (Ruth, 1986). The REL and PEL for hydrogen sulphide are  $15,000 \,\mu g/m^3$  ceiling (10 min) and  $2.77 \times 10^7 \,\mu g/m^3$  ceiling, respectively (OSHA, 1993; NIOSH, 2004). Ceiling concentration limits are the highest concentrations allowable for human exposure. In the case of hydrogen sulphide, the REL and PEL are almost eight orders of magnitudes higher than the OTC. The EPA Region 9 ambient air PRG for hydrogen sulphide is 1  $\mu g/m^3$  (EPA, 2004).

Similar to hydrogen sulphide, the REL and PEL for carbon disulphide are between three to six orders of magnitudes higher than its OTC. Carbon disulphide has an OTC of 24.3  $\mu$ g/m<sup>3</sup> (Ruth, 1986). Carbon disulphide's REL and PEL are 3,000  $\mu$ g/m<sup>3</sup> and 6.23 × 10<sup>7</sup>  $\mu$ g/m<sup>3</sup>, respectively (OSHA, 1997; NIOSH, 2004). For this compound, the EPA Region 9 ambient air PRG is 730  $\mu$ g/m<sup>3</sup>. Aside from being sulphur compounds, the

Compounds	Formula	Formula Odour		отс		B.P. (°C)	M.W.	Ref.	NIOSH REL (based on 8-hr TWA)	
			μ <b>g/m</b> ³	ppb					μg/m³	ppb
Sulphur compounds - cor	npost wheel; sulphur/cabbage/g	arlic								
Hydrogen sulphide	H <sub>2</sub> S	Rotten eggs	0.7	0.502	3	- 60.3	34.1	9	-	-
Methyl mercaptan	(CH <sub>3</sub> )SH	Sulphidy	0.04	0.02	3	5.9	48.1	9	-	-
Carbon oxysulphide	COS	Pungent	-	-	-	-50.2	60.1	9	-	-
Dimethyl sulphide	CH <sub>3</sub> SCH <sub>3</sub>	Decayed cabbage	2.5	0.984	3	37.3	62.1	9	-	-
Ethyl mercaptan	CH <sub>3</sub> CH <sub>2</sub> SH	Garlic	0.03	0.013	3	35	62.1	9	-	-
Sulphur dioxide	SO <sub>2</sub>	Irritating	1,175	448	3	- 10.1	64.1	9	5,000	2,000
Allyl mercaptan	$CH_2 = CHCH_2$ -SH	Garlic-like	0.2	0.066	3	67.5	74.2	9	-	-
Carbon disulphide	CS <sub>2</sub>	Disagree, sweet	24.3	7.7	3	46.2	76.1	9	3,000	10,000
Propyl mercaptan	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> SH	Unpleasant	0.2	0.064	3	67	76.2	9	_	-
Crotyl mercaptan	$CH_3CH = CHCH_2SH$	Skunk-like	0.37	0.1	3	-	90.2	3	-	-
Dimethyl disulphide	$(CH_3)_2S_2$	Rotten cabbage	0.1	0.026	3	109.7	94.2	9	-	-
Thiophenol	C <sub>6</sub> H <sub>5</sub> SH	Putrid, garlic	1.2	0.266	3	169	110.2	9	-	-
Benzyl mercaptan	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> SH	Unpleasant	13.2	2.6	3	195	124.2	9	-	-
Dimethyl trisulphide	$(CH_3)_2S_3$	Rotten cabbage	6.2	1.2	3	-	126.2	9	-	-
Nitrogen compounds - co	mpost wheel: fishy/ammonia	C								
Ammonia	NH <sub>3</sub>	Pungent, irritating	26.6	38.3	3	-33.4	17.0	9	18,000	25,000
Methylamine	CH <sub>3</sub> NH <sub>2</sub>	Fish, ammonia-like	25	20	3	-6.0	31.1	9	12,000	10,000
Dimethylamine	(CH <sub>3</sub> ) <sub>2</sub> NH	Fishy, ammonical	37.8	20.5	3	6.9	45.1	9	18,000	10,000
Trimethylamine	(CH <sub>3</sub> ) <sub>3</sub> N	Fishy, pungent	0.80	0.332	3	2.9	59.1	9	24,000	10,000
Volatile fatty acids - comp	oost wheel: rancid									
Formic acid	НСООН	Pungent, penetrating	45	24	3	100.7	46.0	9	9,000	5,000
Acetic acid	CH <sub>3</sub> COOH	Sour, vinegar-like	2,500	1,017	3	117.9	60.1	9	25,000	10,000
Propionic acid	CH <sub>3</sub> CH <sub>2</sub> COOH	Sour	84	28	3	140.7	74.1	9	30,000	10,000
Butyric acid	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> COOH	Sour, perspiration	1.00	0.278	3	163.5	88.1	9	,     –	· _
Valeric acid	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> COOH	Unpleasant	2.60	0.624	3	186.0	102.1	9	-	-
Caprinic acid	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>8</sub> COOH	Rancid, sour	11,951	1,696	3	269.0	172.3	9	-	-
	post wheel: fragrant/fruity and s		, · · ·	,						
Acetaldehyde	CH <sub>3</sub> CHO	Green sweet, fruity	0.2	0.111	3	20.1	44.1	9	180,000	100,000
Formaldehyde	CH <sub>2</sub> O	Unpleasant	1,470	1198	3	- 19.5	30.0	9	_	16

Table 1A Human detection limits and regulatory limits of odorous gases associated with wastewater treatment and compost

## Table 1A (continued)

Compounds	Formula	Odour OTC		тс	Ref.	B.P. (°C)	M.W.	Ref.	NIOSH REL (based on 8-hr TWA)	
			μg/m³	ррb					μg/m³	ppb
Acrolein	CH2CHCHO	Burnt, sweet	52	22.6	3	52.7	56.1	9	250	100
Propionaldehyde	CH <sub>3</sub> CH <sub>2</sub> CHO	Sweet, ester	22	9.3	З	49	58.1	9	-	-
Acetone	CH <sub>3</sub> COCH <sub>3</sub>	Sweet, minty	47,466	20,016	З	56.2	58.1	9	590,000	250,000
Crotonaldehyde	CH₃CHCHCHO	Pungent, suffocating	105	36.7	3	102	70.1	9	6,000	2,000
Butanaldehyde	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CHO	Sweet	28,000	9495	З	76	72.1	9	-	-
Butanone	CH <sub>3</sub> COCH <sub>2</sub> CH <sub>3</sub>	Sweet, minty	738	251	З	79.6	72.1	9	590,000	200,000
Valeraldehyde	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> CHO	Pungent	98	27.8	З	103	86.1	9	175,000	50,000
2-Pentanone	CH <sub>3</sub> COCH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	Sweet	28,000	7,967	2	105	86.1	9	530,000	150,000
2,4-Heptadienal	CH <sub>3</sub> CH <sub>2</sub> (CH) <sub>5</sub> O	-	-	-	-	84	110.0	9		-
2,4-Decadienal	$CH_3(CH_2)_4(CH)_5O$	-	-	-	-	-	152.0	9	-	-
1-Dodecanal	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>10</sub> CHO	Clean, fresh	15.2	2.02	3	185	184.3	9	-	_
Substituted benzenes - com	post wheel: solventy/hydrocarbo	on								
Benzene	C <sub>6</sub> H <sub>6</sub>	Sweet, solventy	4,500	1,409	3	80	78.1	9	319	100
Toluene	$C_6H_5CH_3$	Rubbery, mothballs	8,025.00	2,130.40	З	111.1	92.1	9	375,000	100,000
Methyl methacrylate	$CH_2 = C(CH_3)COOHCH_3$	Arid, fruity, sulphidy	205	50	3	101.1	100.1	9	410,000	100,000
Styrene	$C_6H_5CH = CH_2$	Solventy, rubbery	202.1	47.4	3	145	104.2	9	215,000	50,000
Ethyl benzene	$CH_3CH_2C_6H_5$	Aromatic	8,700.00	2,003.00	З	136.2	106.2	9	435,000	100,000
m Xylene	$C_6H_4(CH_3)_2$	Sweet	348	80.12	З	138.9	106.2	9	435,000	100,000
p Xylene	$C_6H_4(CH_3)_2$	Sweet	348	80.12	З	138.3	106.2	9	435,000	100,000
1,2,4-trimethylbenzene	$C_6H_3(CH_3)_3$	Distinctive aromatic	-	-	-	169.4	120.2	9	125,000	25,000
1,3,5-trimethylbenzene	$C_6H_3(CH_3)_3$	Distinctive aromatic	-	-	-	165	120.2	9	125,000	25,000
n-Propyl benzene	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	-	-	-	-	159	120.2	9	-	-
Cumene (isopropylbenzene)	$C_6H_5CH(CH_3)_2$	Sharp, aromatic	39.2	8	З	152.2	120.2	9	245,000	50,000
Naphthalene	C <sub>10</sub> H <sub>8</sub>	Mothball, tar-like	1,500	286	З	217.8	128.2	9	50,000	10,000
<i>n</i> -Butyl benzene	$C_6H_5CH_2CH_2CH_2CH_3$	_	-	-	-	183	134.2	9	-	-
<i>p</i> -lsopropyl toluene	$C_6H_4CH_3(CH(CH_3)_2)$	Fresh, citrus	-	-	-	176	134.2	9	-	-
1,4-Dichlorobenzene	C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub>	Mothballs	90,000	14,969	З	173.9	147.0	9	450,000	75,000
Complex N compounds I - co			-						·	
Indole	Ċ <sub>8</sub> H <sub>7</sub> N	Strong, moth ball	-	-	_	253	117.2	9	-	-
Skatole	C <sub>6</sub> H <sub>5</sub> C(CH <sub>3</sub> )CHNH	Perfume	$4.0 \times 10^{-4}$	$7.48 \times 10^{-5}$	3	265	131.1	9	_	_

## Table 1A (continued)

Compounds	Formula	Odour	c	отс			M.W.	Ref.	NIOSH REL (based o 8-hr TWA)	
			μg/m³	ppb					μg/m³	ppb
Complex N compounds I	I - compost wheel: putrid/dead ar	nimal								
Pyridine	C <sub>5</sub> H <sub>5</sub> N	Burnt, sickening	9	2.8	3	115.6	79.1	9	15,000	5,000
Putrescine	NH <sub>2</sub> (CH <sub>2</sub> ) <sub>4</sub> NH <sub>2</sub>	Putrid, rotting flesh	-	-	-	158	88.2	9	· _	-
Cadaverine	NH <sub>2</sub> (CH <sub>2</sub> ) <sub>5</sub> NH <sub>2</sub>	-	-	-		179	102.2	9	-	-
Complex alcohols - com	post wheel: earthy/musty/mouldy									
2-Methylisoborneol	C <sub>11</sub> H <sub>20</sub> O	-	-	-	-	-	168.2	9	-	-
Geosmin	$C_{12}H_{22}O$	Earthy	-	-	-	-	182.3	9	-	-
2,4,6 Trichloranisole	C <sub>6</sub> H <sub>2</sub> (Cl) <sub>3</sub> OCH <sub>3</sub>	-	-	-	-	265.1	211.5	9	-	-
Complex fragrances - co	ompost wheel: terpenes/pine/lemo	n								
Alpha-Pinene	C <sub>10</sub> H <sub>16</sub>	Sweet, pine	-	-	-	155	136.2	9	-	-
Beta-Pinene	$C_{10}H_{16}$	_	-	-	-		136.2	9	-	-
D-Limonene	C <sub>6</sub> H <sub>7</sub> CH <sub>3</sub> CCCH <sub>3</sub>	Lemon, sweet	-	-	-	176	136.2	9	-	-
Eucalyptol	C <sub>10</sub> H <sub>18</sub> O	, _	-	-	-	176	154.3	9	-	-
Menthol	C <sub>6</sub> H <sub>9</sub> CH(CH <sub>3</sub> ) <sub>2</sub> CH <sub>3</sub> OH	Pungent	-	-	-	212	156.3	9	-	-
Vanillin	C <sub>6</sub> H <sub>3</sub> OH(OCH <sub>3</sub> )CHO	Perfume	$2.0 \times 10^{-4}$	$3.2 \times 10^{-5}$	3	285	152.1	9	-	-
Other compounds - con	npost wheel: grassy/woody/smoky									
Cis-3-Hexen-1-ol	CH <sub>3</sub> CH <sub>2</sub> CHCHCH <sub>2</sub>									
	CH <sub>2</sub> OH	Sweet, alcohol	-	-	_	156	100.2	9	-	_
Cis-3-Hexyl acetate		,								
,	COOCH3	Unpleasant	12.00	2.030	3	168	144.2	9	300,000	50,000
Other compounds - not	on compost odour wheel	I								, ,
Phenol	C <sub>6</sub> H <sub>5</sub> OH	Medicinal, sweet	178.6	46.501	3	181.7	94.1	9	-	-
Heptanol	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>6</sub> OH	_	14	2.95	1	175.8	116.2	9	_	_
Benzothiazole	C <sub>6</sub> H <sub>4</sub> SCHN	Penetrating	442	80	3	231	135.1	9	_	_

Table 1B Regulatory limits of odorous gasses associated with wastewater treatment and compost

Compounds	NIOSH STEL (based	NIOSH STEL (based on 15-min TWA)		ef. OSHA PEL (based on 8-hour TWA)		Ref.	Region 9 PRG	Ref.	WHO	Ref.
	μg/m³	ppb		μg/m³	ppb		µg/m³		μ <b>g/m</b> ³	
Sulphur compounds - com	post wheel: sulphur/ca	bbage/garlic								
Hydrogen sulphide	15,000 ceiling	10,000 ceiling	7	$2.77 \times 10^7$ ceiling	20,000 ceiling	6	1	4	150	10
Methyl mercaptan	1000	500 ceiling	7	20,000 ceiling	10,000 ceiling	5	2.1	4	-	-
Carbon oxysulphide	-	_	-	-	_	-	-	-	-	-
Dimethyl sulphide	-	-	-	-	-	-	-	-	-	-
Ethyl mercaptan	1,300	500 ceiling	7	25,000 ceiling	10,000 ceiling	5	-	-	-	-
Sulphur dioxide	13,000	5,000	7	13,000	5,000	5	-	-	125	10
Allyl mercaptan	_	_	-	_	_	-	-	-	-	-
Carbon disulphide	30,000	10,000	7	$6.23 \times 10^7$ ceiling	20,000 (30,000 ceiling)	6	730	4	100	10
Propyl mercaptan	1,600	500	7	-	_	-	-	-	-	-
Crotyl mercaptan	-	-	-	-	_	-	-	-	-	-
Dimethyl disulphide	-	-	-	-	_	-	-	-	-	-
Thiophenol	500	100 ceiling	7	-	_	-	-	-	-	-
Benzyl mercaptan	-	-	-	-	_	-	-	-	-	-
Dimethyl trisulphide	-	-	-	-	_	-	-	-	-	-
Nitrogen compounds - cor	npost wheel: fishy/amm	nonia								
Ammonia	27,000	35,000	7	35,000	50,000	5	100	4	-	-
Methylamine	-	-	7	12,000	10,000	5	-	-	-	-
Dimethylamine	-	-	7	18,000	10,000	5	0.021	4	-	_
Trimethylamine	36,000	15,000	7	-	, _	-	-	_	-	_
Volatile fatty acids - compo	ost wheel: rancid									
Formic acid	-	-	7	9,000	5,000	5	3.1	4	-	-
Acetic acid	37,000	15,000	7	25,000	10,000	5	-	-	-	-
Propionic acid	45,000	15,000	7	-	_	-	-	-	-	-
Butyric acid	-	-	-	-	_	-	-	-	-	-
Valeric acid	-	-	-	-	_	-	-	-	-	-
Caprinic acid	-	-	_	-	_	-	-	-	-	_
Aldehydes/ketones - comp	oost wheel: fragrant/frui	ty and sweet								
Acetaldehyde	270,000	150,000	7	360,000	200,000	5	0.87	4	-	-
Formaldehyde	. –	100	7	-	-	-	0.15	4	100	10
Acrolein	-	-	7	-	_	_	0.021	4	-	_

#### Table 1B (continued)

Compounds	NIOSH STEL (ba	sed on 15-min TWA)	Ref.	OSHA PEL (based o	n 8-hour TWA)	Ref.	Region 9 PRG	Ref.	wнo	Ref.
	μ <b>g/m</b> ³	ppb		μg/m³	ppb		µg/m³		μg/m³	
Propionaldehyde	_	_	_	_	-	_	_	-	-	
Acetone	-	-	7	2,400,000	1,000,000	5	3,300	4	-	-
Crotonaldehyde	-	-	7	-	-	-	0.0035	4	-	-
Butanaldehyde	-	-	-	-	-	-	-	-	-	-
Butanone	885,000	300,000	7	590,000	200,000	5	-	-	-	-
Valeraldehyde	-	-	7	-	-	-	-	-	-	-
2-Pentanone	-	-	7	700,000	200,000	5	-	-	-	-
2,4-Heptadienal	-	-	-	-	-	-	-	-	-	-
2,4-Decadienal	-	-	-	-	-	-	-	-	-	-
1-Dodecanal	-	-	-	-	-	-	-	-	-	-
Substituted benzenes - comp	ost wheel: solven	ity/hydrocarbon								
Benzene	3,190	1,000	7	3,190	1,000	6	0.25	4	-	-
Toluene	560,000	150,000	7	753,600	200,000	6	400	4	0.26	10
Methyl methacrylate	_	_	7	410,000	100,000	5	-	-	-	-
Styrene	425,000	100,000	7	426,000	100,000	6	1,100	4	0.26	10
Ethyl benzene	545,000	125,000	7	435,000	100,000	5	1,100	4	-	-
<i>m</i> -Xylene	655,000	150,000	7	435,000	100,000	5	110	4	-	-
<i>p</i> -Xylene	655,000	150,000	7	435,000	100,000	5	110	4	-	-
1,2,4-Trimethylbenzene	_	_	-	-	_	-	6.2	4	-	-
1,3,5-Trimethylbenzene	-	-	-	-	-	-	6.2	4	-	-
n-Propyl benzene	-	-	-	-	-	-	-	-	-	-
Cumene (isopropylbenzene)	-	-	7	245,000	50,000	5	400	4	-	-
Naphthalene	75,000	15,000	8	50,000	10,000	5	-	-	-	-
<i>n</i> -Butyl benzene	-	-	-	-	-	-			-	-
<i>p</i> -lsopropyl toluene	-	-	-	-	-	-	-	-	-	-
1,4-Dichlorobenzene	-	-	7	450,000	75,000	5	-	-	-	-
Complex N compounds I - co	mpost wheel – fa	aecal/sewery								
Indole	-	-	-	-	-	-	-	-	-	-
Skatole	-	-	-	-	-	-	-	-	-	_
Complex N compounds II - co	mpost wheel - r	outrid/dead animal								
Pyridine	. – '	-	7	15,000	5,000	5	3.7	4	_	_

#### Table 1B (continued)

Compounds	NIOSH STEL (b	ased on 15-min TWA)	Ref.	OSHA PEL (based of	on 8-hour TWA)	Ref.	Region 9 PRG	Ref.	wно	Ref.
	μg/m³	ppb		μ <b>g/m</b> ³	ppb		μg/m³		µg/m³	
Putrescine	_	_	-	_	_	-	_	_	-	_
Cadaverine	-	-	-	-	-	-	-	-	-	-
Complex alcohols - compo	st wheel: earthy/mu	isty/mouldy								
2-Methylisoborneol	_	-	-	-	-	-	-	-	-	-
Geosmin	-	-	-	-	-	-	-	-	-	-
2,4,6-Trichloranisol	-	-	-	-	-	-	-	-	-	-
Complex fragrances - com	post wheel: terpene	es/pine/lemon								
Alpha-Pinene	_	-	-	-	-	-	-	-	-	-
Beta-Pinene	-	-	-	-	-	-	-	-	-	-
D-Limonene	-	-	-	-	-	-	-	-	-	-
Eucalyptol	-	-	-	-	-	-	-	-	-	-
Menthol	-	-	-	-	-	-	-	-	-	-
Vanillin	-	-	-	-	-	-	-	-	-	-
Other compounds - compo	ost wheel: grassy/w	oody/smoky								
Cis-3-Hexen-1-ol	-	-	-	-	-	-	-	-	-	-
Cis-3-Hexyl acetate	-	-	7	300,000	50,000	5	-	-	-	-
Other compounds - not on	compost odour wh	neel								
Phenol	19,000	5,000	7	19,000	5,000	5	1,100	4	-	-
Heptanol	-	-	-	-	-	-	-	-	-	-
Benzothiazole	-	-	-	-	-	-	-	-	-	-

[1] Nagy, 1991. 50% odour detection limit

[2] Miller, 1993. Low human detection limit

[3] Ruth, 1986. Low odour detection limit

[4] EPA Region 9 PRG Table, October 2004

[5] OSHA, Table Z-1 Limits for Air Contaminants, 1993

[6] OSHA, Table Z-2 Limits for Air Contaminants, 1997

[7] NIOSH, Pocket Guide to Chemical Hazards, 2004

[8] NIOSH, Pocket Guide to Chemical Hazards, Appendix G, 1989 Air Contaminants Project, Exposure Limits Not in Effect

[9] The Merck Index, 13th edition, 2001

[10] World Health Organization, Air Quality Guidelines, 2nd Edition, 2000

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characteristic that links these two compounds together is that they have OTCs below their REL, PEL and PRG concentrations.

Unlike hydrogen sulphide and carbon disulphide, sulphur dioxide becomes hazardous at concentrations closer to or below its OTC. The REL and PEL of sulphur dioxide, which has no strong pungent smell attached to its presence, are only one order of magnitude greater than its OTC (Table 1A and B). The OTC of sulphur dioxide is  $1,175 \,\mu g/m^3$  (Ruth, 1986) while the REL and PEL are  $5,000 \,\mu g/m^3$  and  $13,000 \,\mu g/m^3$ , respectively (OSHA, 1993; NIOSH, 2004). According to WHO, hazardous concentrations begin at  $125 \,\mu g/m^3$  (WHO, 2000). Sulphur dioxide is one of several criteria pollutants in our air today. National Ambient Air Quality Standards (NAAQS) require sulphur dioxide concentrations not to exceed an annual concentration of  $80 \,\mu g/m^3$ . Clearly, before sulphur dioxide can be humanly detected, it could already pose a health hazard. This is in contrast to carbon disulphide and hydrogen sulphide, which have strong odours detectable at very low concentrations with RELs and PELs between six and seven magnitudes greater than their OTCs.

#### Nitrogen compounds: fishy/ammonia

Table 1A and B also indicate the present knowledge regarding nitrogen (N) compounds that could be part of the compost plant's odour quality. As with sulphur compounds, the relationship between the many N-type chemicals present, their relative concentrations and the type of odours has not yet clearly been defined. Ammonia and trimethylamine comprise most of the odorous N emissions from wastewater, composting and organic biomass decay. Ammonia produces a pungent medicinal odour, while trimethylamine produces a fishy odour with a human detection limit 100 times lower than ammonia (Rosenfeld and Henry, 2001; Rosenfeld *et al.*, 2002).

Table 1B illustrates that compounds producing a fishy/ammonia-like odour have OTCs well below their REL. For instance, ammonia has an OTC of 26.6  $\mu$ g/m<sup>3</sup> while its REL is 18,000  $\mu$ g/m<sup>3</sup>. Dimethylamine has an OTC which is below the REL and PEL but above the PRG. This indicates a possibility of hazardous conditions before the OTC is reached. Dimethylamine has an OTC of 37.80  $\mu$ g/m<sup>3</sup> (Ruth, 1986). Both the REL and PEL for dimethylamine are 18,000  $\mu$ g/m<sup>3</sup> while the Region 9 EPA ambient air PRG is 0.021  $\mu$ g/m<sup>3</sup>. The PRG is three orders of magnitudes lower than the OTC (EPA, 2004).

#### Volatile fatty acid: rancid

Compounds such as volatile fatty acids are thought to be present in most compost treatment plants that could be part of the compost plant's odour quality. The relationship between these compounds, their relative concentrations and the type of odours has also not been defined. Aerobic secondary treatment produces oxygenated compounds, such as aldehydes, alcohols, ketones and volatile fatty acids (Mosier *et al.*, 1977). Anaerobic digestion processes during compost treatment result in the formation of volatile fatty acids (Killham, 1994). Volatile fatty acids have a rancid, vinegar and body odour-like smell (see Table 1A and B) (Rosenfeld and Suffet, 2004). The volatile fatty acids listed in Table 1B also have OTCs which are below their REL. However, formic acid has an OTC which is below the PEL and REL but above the PRG. The OTC of formic acid is  $45.00 \,\mu g/m^3$  (Ruth, 1986), while its REL and PEL are two magnitudes higher at  $9,000 \,\mu g/m^3$  (OSHA, 1993; NIOSH, 2004). The Region 9 USEPA ambient air PRG for formic acid is  $3.1 \,\mu g/m^3$ , much lower than the compound's REL and PEL (EPA, 2004).

## Aldehydes and ketones: fragrant/fruity and sweet

Aldehydes and ketones generally have sweet pungent odours that result from incomplete decomposition of organic matter during biosolids production. While the sweet solvent-like odours of ketones and aldehydes may not be perceived as unpleasant, mixed with other odorants they contribute to a generally unpleasant odour (Rosenfeld and Suffet, 2004; Rosenfeld *et al.*, 2004).

Acetaldeyde is similar to hydrogen sulphide and carbon disulphide, in that it gives exposed individuals a warning because the OTC is far below the REL, PEL and PRG concentrations. Acetaldehyde has a REL and PEL five orders of magnitudes higher than its OTC at 0.20  $\mu$ g/m<sup>3</sup> (Ruth, 1986; OSHA, 1993; NIOSH, 2004). The EPA Region 9 ambient air PRG for acetaldehyde is 0.87  $\mu$ g/m<sup>3</sup> (EPA, 2004). Acetone may potentially present hazardous conditions before the OTC is reached. The OTC of acetone is below the PEL and REL, but above the EPA Region 9 PRG values. Acetone has an OTC of 47,466  $\mu$ g/m<sup>3</sup> (Ruth, 1986). Acetone's REL and PEL, 590,000  $\mu$ g/m<sup>3</sup> and 2.4 × 10<sup>6</sup>  $\mu$ g/m<sup>3</sup>, respectively, are both between one and two orders of magnitude greater than their OTCs (OSHA, 1993; NIOSH, 2004). However, EPA Region 9 ambient air PRG for acetone is 3,300  $\mu$ g/m<sup>3</sup>, which is one order of magnitude lower than the OTC (EPA, 2004), indicating the potential for hazardous conditions before humans detect the compounds' odour.

#### Complex N-compounds: faecal/ sewery and putrid/dead animal

Composting can also generate complex nitrogen compounds which produce a faecal and sewery odour or with a putrid and dead animal-like odour. More research needs to be conducted in this area of odour to determine the OTCs for these compounds. Pyridine, however, produces a burnt, sickening odour which is associated with a dead animal. For this compound, the OTC of  $9 \,\mu g/m^3$  is well below the REL of  $15,000 \,\mu g/m^3$ .

### Complex alcohols and fragrances: earthy/musty/mouldy and terpenes/pine/lemon

Complex alcohols and fragrances are often noted at composting facilities. These produce an earthy, musty odour or a pine and lemon odour. However, more research needs to be conducted in this specific odour area to establish OTCs, RELs and PELs.

#### Other compounds: grassy/woody/smoky

A grassy, woody, smoky odour is also often detected near compost facilities. The unpleasant, grassy odour of *cis*-3-hexyl acetate is noticed well below its REL. The OTC of this compound is  $12 \,\mu g/m^3$  while its REL is  $300,000 \,\mu g/m^3$ . Essentially, there is no definitive relationship between strong, pungent smells and hazardous concentrations. Instead, these comparisons demonstrate that compounds with high OTCs can pose a greater risk than compounds that have low OTCs.

## Substituted benzenes: solventy/hydrocarbon

Komilis *et al.* (2004) identified various xenobiotic VOCs in the gaseous emissions of organic municipal solid waste components. The VOCs appear to be embedded in the solid matrix and are released upon wetting and heating during the initiation of the composting process. The results demonstrated that municipal solid waste composting can be a source of hazardous VOC emissions.

Several substituted benzenes have OTCs which are below the REL and PEL but above the PRG. Ethyl benzene has an OTC of  $8,700 \,\mu g/m^3$ . The RELs and PELs for ethyl benzene is at least two orders of magnitude higher than its OTC. The EPA Region 9 ambient air PRG for ethyl benzene is  $1,100 \,\mu g/m^3$ , once again much lower than its OTC (EPA, 2004). In addition, toluene has a PRG of  $400 \,\mu g/m^3$  (EPA, 2004), much lower

than its OTC at  $8,025 \,\mu g/m^3$  (Ruth, 1986). Cumene (isopropylbenzene) gives a warning to an exposed individual by having an OTC that is below its REL and PEL, but cumene does not have a PRG value. Cumene's OTC is  $39.20 \,\mu g/m^3$  and both the REL and the PEL are  $245,000 \,\mu g/m^3$  (Ruth, 1986; OSHA, 1993; NIOSH, 2004).

Benzene is a case in which the REL PEL and PRG are lower than the OTC. Therefore, there is no warning smell associated with hazardous benzene exposure. Benzene is a known carcinogen and has an OTC of  $4,500 \,\mu g/m^3$ . The REL and PEL for benzene are  $319 \,\mu g/m^3$  and  $3,190 \,\mu g/m^3$ , which are below the OTC (OSHA, 1993; NIOSH, 2004). The EPA Region 9 ambient air PRG of benzene is  $0.25 \,\mu g/m^3$ , which is four orders of magnitude lower then the OTC (EPA, 2004). Benzene is the most hazardous VOC associated with composting and should be monitored.

#### Conclusion

Each odorant possesses a characteristic odour signature-odour quality and odour threshold as well as a toxicity value. Odorants are released into air during the decomposition of organic waste at the compost treatment plant. Composting releases volatile organic chemicals, including alcohols, aldehydes, volatile fatty acids, ammonia and other nitrogen compounds, xenobiotic solvents and various sulphur compounds into the environment. This paper begins to develop an understanding of how the VOCs present in the raw compost (including xenobiotic solvents) are related to human health threshold criteria developed by NIOSH, OSHA, USEPA and WHO. This comparison indicates that: (1) the human OTC for most compost odorants are far lower than their respective human health risk (regulatory) threshold values; (2) several compost odorants have OTC that are below some of their respective regulatory thresholds and above others (i.e. dimethyl amine, formic acid, acetone, ethyl benzene and toluene); (3) only the VOCs probably present as contaminants in the raw composting material have OTC greater than all of its regulatory thresholds (i.e. benzene). Benzene is the most hazardous VOC associated with composting and should be monitored.

#### References

- Banwart, W.L. and Bremner, J.M. (1975). Formation of volatile sulfur compounds by microbial decomposition of sulfur-containing amino acids in soils. *Soil Biol. Biochem.*, 7, 359–364.
- Killham, K. (1994). Soil Ecology, Cambridge University Press, Cambridge, UK, pp. 42.
- Komilis, D.P., Ham, R.K. and Park, J.K. (2004). Emission of volatile organic compounds during composting of municipal solid waste. *Wat. Res.*, 38, 1707–1714.
- Merck Index (2001). An Encyclopedia of Chemicals, Drugs and Biologicals, 13th edn, Merck Research Laboratories. Merck & Co. Inc, Whitehouse Station, NJ, USA.
- Miller, F.C. (1993). Minimizing odor generation. In Science and Engineering of Compost Design, Environmental, Microbiological and Utilization Aspects, Hoitink, H.A.J. and Keener, H.M. (eds), Renaissance Publications, Worthington, OH, USA, pp. 219–241.
- Mosier, A.R., Morrison, S.M. and Elmond, G.K. (1977). Odors and emissions from organic wastes. In *Soil for Management of Organic Waste and Waste Waters*, Soil Sci. Society of America, Madison, WI, USA, pp. 532–569.
- Nagy, G.Z. (1991). The odor impact model. J. Air Waste Management Assoc., 41(10), 1360-1362.
- NIOSH (2004). *Pocket Guide to Chemical Hazards*, PB97-140, National Institute for Occupational Safety and Health, USA.
- OSHA (1997). *Table Z-2 Limit for Air Contaminants*, 1910.1000, Occupational Safety and Health Administration, USA.
- OSHA (1993). *Table Z-1 Limits for Air Contaminants*, 1910.1000, Occupational Safety and Health Administration, USA.

- Rosenfeld, P., Clark, J.J. and Suffet, I.H. (2004). The value of an odor-quality-wheel classification scheme for compost facility evaluations. *Presented at the Biocycle Conference, Los Angeles, CA, USA*.
- Rosenfeld, P.E. and Henry, C.L. (2001). Activated carbon and wood ash sorption of wastewater. *Compost Biosolids Odorants Wat. Env. Res.*, 73, 388–392.
- Rosenfeld, P.E. and Suffet, I.H. (2004). Understanding odorants associated with compost, biomass facilities, and the land application of biosolids. *Wat. Sci. Tech.*, 49(9), 193–199.
- Rosenfeld, P.E., Grey, M. and Suffet, I.H. (2002). Compost Demonstration Project, Sacramento, California Using High-Carbon Wood Ash to Control Odor at a Green Materials Composting Facility Integrated Waste Management Board Public Affairs Office, Publication Clearinghouse (MS-6), Sacramento, CA Publication #442-02-008. April.
- Rosenfeld, P.E., Grey, M.A. and Suffet, M. (2004). Compost odour control using high carbon wood ash. *Wat. Sci. Tech.*, **49**(9), 171–178.
- Ruth, J.H. (1986). Odor thresholds and irritation levels of several chemical substances: a review. Am. Ind. Hyg. Assoc., 47, 142–151.
- Suffet, I.H. and Rosenfeld, P.E. (2007). The anatomy of odour wheels for odours of drinking water, wastewater, compost and the urban environment. *Wat. Sci. Tech.*, 55(5), 335–344.
- US Environmental Protection Agency (EPA) (2004). *Region 9 PRG Table*, US Environmental Protection Agency, Washington, DC, USA.
- WEF (1995). *Odor Control in Wastewater Treatment Plants*, WEF Manual of Practice 22. Water Environmental Federation, Alexandria, VA, USA.
- WHO (2000). Air Quality Guidelines, 2nd edn, No. 91. World Health Organization, Regional Publications, European Series, Geneva, Switzerland.